

Chemistry 3202
Outline and Evaluation
2011-2012
Ms. Ruth Simmons

Unit		Suggested Completion Date
1	Reaction Rates	October 29
2	Acids and Bases	February 24
3	Thermochemistry	April 20
4	Electrochemistry	May 25

Task	Percentage	Value
Tests	35	17.5
Labs and STSEs	15	7.5
Assignments/Quizzes	20	10
Midyear Exam	15	7.5
Pre-public Exam	15	7.5
Government Public Exam	----	50
Total	100	100

Final Grade on transcript in August is equal to: $\frac{\text{School Mark} + \text{Public Exam Mark}}{2}$

Keep in mind your mark may be D-factored up or down based on provincial results

Expectations for Chemistry 3201:

1. This is a fast-paced course that involves a great deal of work. Homework will be on-going to practice concepts introduced during the day. Very little time will be provided to do assignment work during class time, however extra help is available after school. **THIS COURSE IS MUCH MORE DIFFICULT THAN CHEMISTRY 2202.**
2. Menihek's "Second Chances" policy will apply to major evaluation pieces, provided the student is willing to put in the extra work and time required to improve the mark they want to replace. All second chances must be requested in a timely manner.
3. Students are **RESPONSIBLE** for work that is missed when they are absent, for any reason. The student is required to find out what was missed on his/her time from a classmate, do any required work and submit in a timely fashion.

