Completion of Missing Reactants/Products(Review)

1. $\rightarrow$ PbO(s) + Na(s)

2 $\rightarrow$ XeF_6(s)

3. $\rightarrow$ BaO(g)

4. $\rightarrow$ NaOH(s)

5 $\rightarrow$ NaF(s)

6. Al(s) + $Cu(NO_3)_2$(aq)$\rightarrow$

7. I_2(g) + NaBr(aq) $\rightarrow$

8. Fe(s) + H_2S(aq) $\rightarrow$

9. $CaCO_3$(s) + LiOH(aq) $\rightarrow$

10. H_3PO_4(aq) + $CaCl_2$(aq) $\rightarrow$

11. Cr_2O_3(s) $\rightarrow$

12. $\rightarrow$ CuS(s) + $NaCH_3COO$(aq)

13. C_2H_6(g) + O_2(g) $\rightarrow$

14. Na_3BO_3(aq) + $Pb(NO_3)_4$ $\rightarrow$

15. C_7H_{14}(l) + O_2(g) $\rightarrow$

16. I_2(g) + KBr(aq) $\rightarrow$

17. $\rightarrow$ $Na_2C_2O_4$(aq) + $BaCl_2$(aq)

18. $Na(s) + H_2O(l)$ [think HOH not H_2O] $\rightarrow$

19. $V_2(SO_4)_5(s) + Al_2O_3(s) \rightarrow$

20. $\rightarrow CuCr_2O_7(s) + U(NO_3)_3(s)$

21. $C_{20}H_{42}(s) + O_2(g) \rightarrow$

22. $\rightarrow Mn(OH)_2(s) + Mg(HCO_3)_2(aq)$

23. $\rightarrow H_2(g) + NaOH(aq)$

24. $\rightarrow Al(s) + LiBr(s)$

25. $\rightarrow HCl(g) + Na_2SO_3(aq)$

26. $\rightarrow PbO(s) + Na(s)$